#### GOBIERNO DE CENCIA E INNOVACIÓN Y Tecnológicas

## Pore Water Chemistry Lab and Physico-chemistry of materials

	PORE WATER CHEMISTRY LABORATORY	
	PHYSICAL CHARACTERIZATION	<ul> <li>Grain Density</li> <li>Dry Bulk Density</li> <li>Gravimetric Water Content</li> <li>Physical Porosity</li> </ul>
<image/>	MINERALOGICAL CHARACTERIZATION	<ul> <li>X-Ray Diffraction</li> <li>Thermal Analysis, DSC</li> <li>SEM, TEM, XPS, Mössbauer</li> <li>FTIR</li> </ul>
	SURFACE	<ul> <li>RAMAN</li> <li>Total surface area</li> <li>Characterization of the dry state:         <ul> <li>N<sub>2</sub> adsorption isotherms (BET)</li> <li>Porosimetry (Hg Porosimetry)</li> </ul> </li> <li>Charaterization of the saturated state:         <ul> <li>H<sub>2</sub>O adsorption isotherm</li> <li>d(001) basal spacings</li> </ul> </li> </ul>
	CHARACTERIZATION	
	PHYSICO- CHEMICAL PROPERTIES	<ul> <li>Cation exchange population</li> <li>Total CEC</li> <li>Variable Charges</li> <li>Potentiometric titrations</li> <li>Soluble salts by aqueous leaching</li> <li>Pore water extracted by Squeezing</li> </ul>





Ciemat Centro de Investigaciones

Energéticas, Medioar

SOBIERNO DE ESPAÑA

MINISTERIO DE CIENCIA











Centrifuge HETTICH Rotina 380R (11000 rpm)



Jacomex anoxic glove box (< 1 ppm O<sub>2</sub>)



AquaMate 8000 UV-Vis Spectrophotmeter (190-1100 nm)





## Mineralogical, Surface and FQ Characterization

### Water adsorption/desorption isotherms



Vacuum desiccators with sulphuric acid solutions to apply different total suctions



Vacuum desiccators with saturated salts solutions to apply different water pressures



#### Nicolet iS50 FTIR spectrometer:

Spectral range: 11000 – 50 cm<sup>-1</sup> (NIR, MIR, FAR) Diffuse Reflectance Tecnique (DRIFT) Attenuated Total Reflection (ATR, diamond) Coupled to WETSYS water vapour generator



- Purge gas generator:
- CO<sub>2</sub>-free
- -73°C dew point
- No particles > 0.01 μm



TG-DSC-ATD / Water adsorption:

THEMYS 1750 + WETSYS (Water vapour generator): 5-95% RH, 20-70 °C Heat capacity (Cp, J·mol<sup>-1</sup>·K<sup>-1</sup>)



#### GOBIERNO DE ESPAÑA E INNOVACIÓN Y Tecnológicas

## Pore Water Chemistry Lab and Physico-chemistry of materials



Figure 1. a) Hydromechanical analogy for load changes during squeezing and consolidation (Terzaghi, 1925; Lambe and Whitman, 1969): The resistance of the internal phase compression is represented by a spring and the rate at which the pore water flows is dependent on the size of the valve aperture. First the valve is closed and in equilibrium. When a load is applied, the piston load is apportioned by the water and the spring in relation to the stiffness of each. All the applied load is resisted by an increase in the fluid pressure. If the valve is opened, the excess of pore pressure will dissipate by water escaping through the valve. The piston drops and the volume chamber decreases until there is a new equilibrium when the load is carried by the spring and the water pressure has returned to the original hydrostatic condition; b) Illustration of the reduction of volume,  $V_o$  to  $V_f$ , of a saturated core sample from Opalinus Clay (BHT-1 m. 12.42) after a squeezing test. The volume of expelled pore water is represented by  $V_{pw}$ . (Fernández et al., 2014)



# Types of Squeezing cells



## up to 100 MPa

L = 250 mm 20 mm wall,  $\phi_{inner}$ = 70 mm

## up to 200 MPa

L = 500 mm 45 mm wall,  $\phi_{inner}$  = 70 mm up to 800 MPa

L = 250 mm 90 mm wall,  $\phi_{inner}$  = 60 mm



Ciemat

Centro de Investigacione

Energéticas, Medioambi

GOBIERNO DE ESPAÑA MINISTERIO DE CIENCIA

EINNOVACIÓ

## **Applying Squeezing Technique**



#### GOBIERNO DE ESPAÑA E INNOVACIÓN E INNOVACIÓN Centro de rivestigaciones E INNOVACIÓN Y Tecnaidgioas

## Pore Water Chemistry Lab and Physico-chemistry of materials

# Applying Squeezing Technique



Deviation of chloride concentrations at high squeezing pressures (Cl(p)) from those measured at lowest pressure yielding water (Cl(po)) for squeezing tests (Fernández et al., 2014)









# **Global Methodology**

